

Chemeco Spray Clean

Chemwatch Hazard Alert Code: 2

Issue Date: 27/06/2022 Print Date: 05/03/2023 S.GHS.AUS.EN

Chemwatch: 24-8912

Safety Data Sheet according to WHS and ADG requirements

SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Identifier

Version No: 2.1.1.1

Product name	Spray Clean – Spray & Wipe
Synonyms	Not Available
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses General purpose cleaner.

Details of the supplier of the safety data sheet

Registered company name	Chemeco (Aust)
Address	17 Yale Drive Epping VIC 3076
Telephone	+61 3 9408 8699
Fax	+61 3 9408 8399
Website	www.chemeco.com.au
Email	info@chemeco.com.au

Emergency telephone number

Association / Organisation	Not Available
Emergency telephone numbers	Not Available
Other emergency telephone numbers	Not Available

SECTION 2 HAZARDS IDENTIFICATION

Classification of the substance or mixture

NON-HAZARDOUS CHEMICAL. NON-DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

Poisons Schedule	Not Applicable
Classification	Not Applicable
Label elements	
GHS label elements	Not Applicable
SIGNAL WORD	NOT APPLICABLE

Hazard statement(s)

Not Applicable

Precautionary statement(s) Prevention

Not Applicable

Precautionary statement(s) Response

Not Applicable

Precautionary statement(s) Storage

Not Applicable

Precautionary statement(s) Disposal

Not Applicable

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SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
10213-79-3	<10	sodium metasilicate, pentahydrate
111-76-2	<10	ethylene glycol monobutyl ether
Not Available	<10	surfactants
Not Available	<1	perfume
Not Available	<1	dye
7732-18-5	>60	water

SECTION 4 FIRST AID MEASURES

Description of first aid measures

Eye Contact	If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary.
Ingestion	 If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Seek medical advice.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

SECTION 5 FIREFIGHTING MEASURES

Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility	Fire Incompatibility None known.		
Advice for firefighters			
Fire Fighting Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water courses. Use fire fighting procedures suitable for surrounding area.			

Fire/Explosion Hazard

- Non combustible. Not considered to be a significant fire risk.
- Expansion or decomposition on heating may lead to violent rupture of containers.
- Decomposes on heating and may produce toxic fumes of carbon monoxide (CO).

Decomposes on heating and produces toxic fumes of:, carbon dioxide (CO2), nitrogen oxides (NOx), sulfur oxides (SOx), phosphorus oxides (POx)

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Slippery when spilt.

Minor Spills

▶ Clean up all spills immediately.

Avoid breathing vapours and contact with skin and eyes.

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Control personal contact with the substance, by using protective equipment. F Contain and absorb spill with sand, earth, inert material or vermiculite.

Major Spills

Slippery when spilt. Minor hazard.

- · Clear area of personnel.
- Alert Fire Brigade and tell them location and nature of hazard.
- Control personal contact with the substance, by using protective equipment as required.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

Cofe han dilan	 Limit all unnecessary personal contact. Wear protective clothing when risk of exposure occurs.
Safe handling	Use in a well-ventilated area.When handling DO NOT eat, drink or smoke.
Other information	 Store in original containers. Keep containers securely sealed.
Cale illomaton	 Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers.

Conditions for safe storage, including any incompatibilities

- Polyethylene or polypropylene container.
- Packing as recommended by manufacturer.
- Check all containers are clearly labelled and free from leaks

Storage incompatibility

Segregate from

strong acids

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	ethylene glycol monobutyl ether	2-Butoxyethanol	96.9 mg/m3 / 20 ppm	242 mg/m3 / 50 ppm	Not Available	Sk

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
sodium metasilicate, pentahydrate	Sodium metasilicate pentahydrate	45 mg/m3	45 mg/m3	170 mg/m3
sodium metasilicate, pentahydrate	Sodium silicate; (Sodium metasilicate)	18 mg/m3	230 mg/m3	230 mg/m3
ethylene glycol monobutyl ether	Butoxyethanol, 2-; (Glycol ether EB)	20 ppm	20 ppm	700 ppm
In one dient	Original IDLU	Revised IDLH		
Ingredient	Original IDLH	Revised IDLH		
sodium metasilicate,	Not Available	Not Available		

Ingredient	Original IDLH	Revised IDLH
sodium metasilicate, pentahydrate	Not Available	Not Available
ethylene glycol monobutyl ether	700 ppm	700 [Unch] ppm
surfactants	Not Available	Not Available
perfume	Not Available	Not Available
dye	Not Available	Not Available
water	Not Available	Not Available

Exposure controls

Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

Personal protection







Eye and face protection

- Safety glasses with side shields; or as required,
- Chemical goggles
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience

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Skin protection	See Hand protection below	
	Wear chemical protective gloves, e.g. PVC.	
Hands/feet protection	Wear safety footwear.	
Body protection	See Other protection below	
Others and the second	, Overalls.	
Other protection	▶ Eyewash unit.	
Thermal hazards	Not Available	

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computer-generated selection:

NV Chemicals Multi Purpose Cleaner

Material	СРІ
BUTYL	A
NEOPRENE	A
NATURAL RUBBER	С
PVA	С

^{*} CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove,

a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as

Respiratory protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 5 x ES	A-AUS / Class 1 P2	-	A-PAPR-AUS / Class 1 P2
up to 25 x ES	Air-line*	A-2 P2	A-PAPR-2 P2
up to 50 x ES	-	A-3 P2	-
50+ x ES	-	Air-line**	-

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

	ormation on basic physical and one mean			
Appearance	Clear pink fragrant liquid; mixes with water.			
Physical state	Liquid	Relative density (Water = 1)	1.095-1.105	
Odour	Not Available	Partition coefficient n-octanol / water	Not Available	
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available	
pH (as supplied)	Not Available	Decomposition temperature	Not Available	
Melting point / freezing point (°C)	0	Viscosity (cSt)	Not Available	
Initial boiling point and boiling range (°C)	100	Molecular weight (g/mol)	Not Applicable	
Flash point (°C)	Not Available	Taste	Not Available	
Evaporation rate	Not Available	Explosive properties	Not Available	
Flammability	Not Available	Oxidising properties	Not Available	
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Available	
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available	
Vapour pressure (kPa)	2.3 @ 20 C	Gas group	Not Available	
Solubility in water (g/L)	Miscible	pH as a solution (1%)	11.1-11.5	
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available	

SECTION 10 STABILITY AND REACTIVITY

Dan estimitus	0
Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

[&]quot;feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

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SECTION 11 TOXICOLOGICAL INFORMATION

Inhaled	Not normally a hazard due to non-volatile nature of product			
Ingestion	Accidental ingestion of the material may be damaging to the health of the individual. Ingestion may result in nausea, abdominal irritation, pain and vomiting			
Skin Contact	There is some evidence to suggest that this material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing skin condition Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.			
Eye	There is some evidence to suggest that this material can cause eye irritation and damage in some persons.			
Chronic	Prolonged or repeated skin contact may cause drying with cracking, irritation and possible dermatitis following. As with any chemical product, contact with unprotected bare skin; inhalation of vapour, mist or dust in work place atmosphere; or ingestion in any form, should be avoided by observing good occupational work practice.			
NV Chemicals Multi	TOXICITY	IRRITATION		
Purpose Cleaner	Not Available	Not Available		
	TOXICITY	IRRITATION		
sodium metasilicate,	Oral (rat) LD50: 847 mg/kg ^[2]	Skin (human):	250 mg/24h SEVERE	
pentahydrate	Oran (rai) EDSS. 847 Highlig		50 mg/24h SEVERE	
	TOXICITY	IRRITATION		
	dermal (rat) LD50: >2000 mg/kg ^[1]	* [Union Carbio	del	
ethylene glycol monobutyl ether	Inhalation (rat) LC50: 450 ppm/4hr ^[2]	į	00 mg SEVERE	
monopatyr etner	Oral (rat) LD50: 250 mg/kg ^[2]		00 mg/24h-moderate	
			00 mg, open; mild	
	тохісіту	IRRITATION		
water	Oral (rat) LD50: >90000 mg/kg ^[2]	Not Available		
SODIUM METASILICATE, PENTAHYDRATE	The material may be irritating to the eye, with prolonged contact causing inflammation. Repeated or prolonged exposure to irritants may produce conjunctivi Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. The material may produce respiratory tract irritation, and result in damage to the lung including reduced lung function.			
	sodium metasilicate anhydrous:		diagnosis of RADS.	
ETHYLENE GLYCOL MONOBUTYL ETHER		ult in damage to the lung including pronounced inflammation. FMAEs): ene ether (EGPE), ethylene gles ADH-3, which catalyzes the of the aldehydes by aldehyde alglycol ethers. embers range from 739 (EGHE ether (2-butoxyethanol) at 10 decreased number of viable is arent in rats. Teratogenic effects were less than that of other the gastrointestinal tract. Limit of slow. Following absorption, encluding humans, ethylene gly	diagnosis of RADS. Iding reduced lung function. Repeated or prolonged exposure to irritants you butyl ether (EGBE) and ethylene glycol hexyl conversion of their terminal alcohols to aldehydes dehydrogenase produces alkoxyacetic acids, which E) to 3089 mg/kg bw (EGPE), with values increasing oppm or rabbits at 200 ppm during organogenes mplantations per litter. Slight foetoxicity in the form cts were not observed in other species. In monoalkyl ethers of ethylene glycol. Ited information suggests that it is also absorbed ethylene glycol is distributed throughout the body yool is initially metabolised by alcohol.	
	sodium metasilicate anhydrous: The material may produce severe irritation to the eye causin may produce conjunctivitis. For ethylene glycol monoalkyl ethers and their acetates (EG Typical members of this category are ethylene glycol propyle ether (EGHE) and their acetates. EGMAEs are substrates for alcohol dehydrogenase isozyme (which are transient metabolites). Further, rapid conversion are the predominant urinary metabolites of mono substituted Acute Toxicity: Oral LD50 values in rats for all category medecreasing molecular weight. Exposure of pregnant rats to ethylene glycol monobutyl eresulted in maternal toxicity and embryotoxicity including a poorly ossified or unossified skeletal elements was also apparat least one researcher has stated that the reproductive efferer ethylene glycol is quickly and extensively absorbed through through the respiratory tract; dermal absorption is apparently according to total body water. In most mammalian species, in	ult in damage to the lung including pronounced inflammation. For MAEs): ene ether (EGPE), ethylene glasse ADH-3, which catalyzes the of the aldehydes by aldehyde glycol ethers. embers range from 739 (EGHE ether (2-butoxyethanol) at 10 decreased number of viable is arent in rats. Teratogenic effects were less than that of other the gastrointestinal tract. Limit of slow. Following absorption, encluding humans, ethylene glynn animals exposed to high concernic and the state of the stat	diagnosis of RADS. Iding reduced lung function. Repeated or prolonged exposure to irritants you butyl ether (EGBE) and ethylene glycol hexyl conversion of their terminal alcohols to aldehydes dehydrogenase produces alkoxyacetic acids, which E) to 3089 mg/kg bw (EGPE), with values increasing oppm or rabbits at 200 ppm during organogenes mplantations per litter. Slight foetoxicity in the form cts were not observed in other species. In monoalkyl ethers of ethylene glycol. Ited information suggests that it is also absorbed ethylene glycol is distributed throughout the body yool is initially metabolised by alcohol.	
MONOBUTYL ETHER	sodium metasilicate anhydrous: The material may produce severe irritation to the eye causin may produce conjunctivitis. For ethylene glycol monoalkyl ethers and their acetates (EG Typical members of this category are ethylene glycol propyle ether (EGHE) and their acetates. EGMAEs are substrates for alcohol dehydrogenase isozyme (which are transient metabolites). Further, rapid conversion of are the predominant urinary metabolites of mono substituted Acute Toxicity: Oral LD50 values in rats for all category medecreasing molecular weight. Exposure of pregnant rats to ethylene glycol monobutyl eresulted in maternal toxicity and embryotoxicity including a poorly ossified or unossified skeletal elements was also apport at least one researcher has stated that the reproductive effer ethylene glycol: Ethylene glycol is quickly and extensively absorbed through through the respiratory tract; dermal absorption is apparently according to total body water. In most mammalian species, it NOTE: Changes in kidney, liver, spleen and lungs are observed in	ult in damage to the lung including pronounced inflammation. For MAEs): ene ether (EGPE), ethylene glasse ADH-3, which catalyzes the of the aldehydes by aldehyde glycol ethers. embers range from 739 (EGHE ether (2-butoxyethanol) at 10 decreased number of viable is arent in rats. Teratogenic effects were less than that of other the gastrointestinal tract. Limit of slow. Following absorption, encluding humans, ethylene glyn animals exposed to high concessearch.	diagnosis of RADS. Iding reduced lung function. Repeated or prolonged exposure to irritants yould butyl ether (EGBE) and ethylene glycol hexyl conversion of their terminal alcohols to aldehydes dehydrogenase produces alkoxyacetic acids, which E) to 3089 mg/kg bw (EGPE), with values increasing opm or rabbits at 200 ppm during organogenes implantations per litter. Slight foetoxicity in the form of the were not observed in other species. In monoalkyl ethers of ethylene glycol. Ited information suggests that it is also absorbed ethylene glycol is distributed throughout the body yool is initially metabolised by alcohol. Intrations of this substance by all routes. ** ASCC (NZ) Sections of the substance of the product of the substance of the sub	
WATER SODIUM METASILICATE, PENTAHYDRATE & ETHYLENE GLYCOL	sodium metasilicate anhydrous: The material may produce severe irritation to the eye causin may produce conjunctivitis. For ethylene glycol monoalkyl ethers and their acetates (EG Typical members of this category are ethylene glycol propyle ether (EGHE) and their acetates. EGMAEs are substrates for alcohol dehydrogenase isozyme (which are transient metabolites). Further, rapid conversion of are the predominant urinary metabolites of mono substituted Acute Toxicity: Oral LD50 values in rats for all category medecreasing molecular weight. Exposure of pregnant rats to ethylene glycol monobutyl eresulted in maternal toxicity and embryotoxicity including a poorly ossified or unossified skeletal elements was also apparat least one researcher has stated that the reproductive effer ethylene glycol is quickly and extensively absorbed through through the respiratory tract; dermal absorption is apparently according to total body water. In most mammalian species, in NOTE: Changes in kidney, liver, spleen and lungs are observed in No significant acute toxicological data identified in literature set.	ult in damage to the lung including pronounced inflammation. For MAEs): ene ether (EGPE), ethylene glasse ADH-3, which catalyzes the of the aldehydes by aldehyde glycol ethers. embers range from 739 (EGHE ether (2-butoxyethanol) at 10 decreased number of viable is arent in rats. Teratogenic effects were less than that of other the gastrointestinal tract. Limit of slow. Following absorption, encluding humans, ethylene glyn animals exposed to high concessearch.	diagnosis of RADS. Iding reduced lung function. Repeated or prolonged exposure to irritants yould butyl ether (EGBE) and ethylene glycol hexyl conversion of their terminal alcohols to aldehydes dehydrogenase produces alkoxyacetic acids, which E) to 3089 mg/kg bw (EGPE), with values increasing opm or rabbits at 200 ppm during organogenes implantations per litter. Slight foetoxicity in the form of the were not observed in other species. In monoalkyl ethers of ethylene glycol. Ited information suggests that it is also absorbed ethylene glycol is distributed throughout the body yool is initially metabolised by alcohol. Intrations of this substance by all routes. ** ASCC (NZ) Sections of the substance of the product of the substance of the sub	

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Serious Eye	0	STOT - Single Exposure	0
Damage/Irritation	0	3101 - Shigle Exposure	0
Respiratory or Skin	6	OTOT Barradad Francisco	
sensitisation	0	STOT - Repeated Exposure	0
Mutagenicity	○ Aspiration	Hazard	0
		Legend: X	 Data available but does not fill the criteria for classification
		•	– Data required to make classification available
		6	- Data Not Available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
sodium metasilicate, pentahydrate	EC50	96	Crustacea	160mg/L	1
sodium metasilicate, pentahydrate	LC50	96	Fish	180mg/L	1
sodium metasilicate, pentahydrate	EC50	48	Crustacea	1700mg/L	2
sodium metasilicate, pentahydrate	EC50	72	Algae or other aquatic plants	207mg/L	2
ethylene glycol monobutyl ether	EC50	384	Crustacea	51.539mg/L	3
ethylene glycol monobutyl ether	LC50	96	Fish	222.042mg/L	3
ethylene glycol monobutyl ether	EC50	48	Crustacea	164mg/L	2
ethylene glycol monobutyl ether	NOEC	168	Crustacea 56mg/L		2
ethylene glycol monobutyl ether	EC50	96	Algae or other aquatic plants 720mg/L		2
water	EC50	384	Crustacea 199.179mg/L		3
water	EC50	96	Algae or other aquatic plants	8768.874mg/L	3
water	LC50	96	Fish 897.520mg/L		3
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	istence: Water/Soil Persistence: Air	
ethylene glycol monobutyl ether	OW (Half-life = 56 days) LOW (Half-life = 1.37 days)		LOW (Half-life = 1.37 days)
water	LOW		LOW

Bioaccumulative potential

Ingredient	Bioaccumulation	
ethylene glycol monobutyl ether	LOW (BCF = 2.51)	
water	LOW (LogKOW = -1.38)	

Mobility in soil

Ingredient	Mobility
ethylene glycol monobutyl ether	HIGH (KOC = 1)
water	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

Product / Packaging disposal

- Recycle wherever possible or consult manufacturer for recycling options.
- Consult State Land Waste Management Authority for disposal.
- Bury residue in an authorised landfill.
- Recycle containers if possible, or dispose of in an authorised landfill.

SECTION 14 TRANSPORT INFORMATION

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Marine Pollutant NO
HAZCHEM Not Applicable

Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

SODIUM METASILICATE, PENTAHYDRATE(10213-79-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Hazardous Substances Information System - Consolidated Lists

Australia Inventory of Chemical Substances (AICS)

ETHYLENE GLYCOL MONOBUTYL ETHER(111-76-2) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Exposure Standards

Australia Hazardous Substances Information System - Consolidated Lists

Australia Inventory of Chemical Substances (AICS)

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC

Monographs

WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Inventory of Chemical Substances (AICS)

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Y
Canada - NDSL	N (sodium metasilicate, pentahydrate; water; ethylene glycol monobutyl ether)
China - IECSC	Y
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (water)
Korea - KECI	Y
New Zealand - NZIoC	Y
Philippines - PICCS	Υ
USA - TSCA	Y
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 OTHER INFORMATION

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

 ${\sf PC-TWA: Permissible \ Concentration-Time \ Weighted \ Average}$

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit $_{\circ}$

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index